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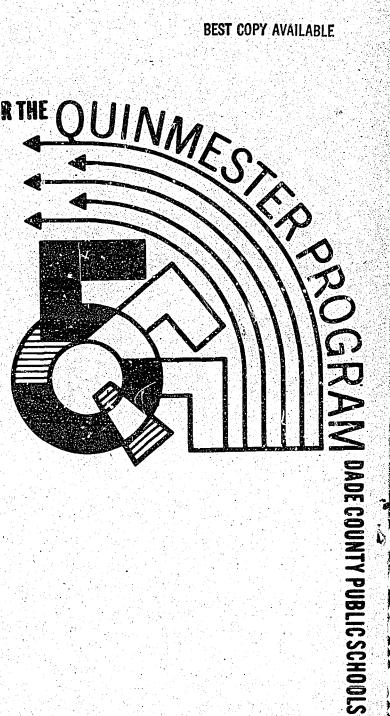
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ABSTRACT

For information on the Visual Arts Education Curriculum of which this is one course, see SO 007 721. In this course students study ceramic material, form, and deconation, historically and technically, and demonstrate competencies in preparing clay bodies and ceramic glazes. Course content is outlined in those areas. Special sections of the guide include lists of the following criteria for evaluating the student's art experience; recipes for glazes, resources such as books and periodicals on history and technique; films and slides and their sources; Professional journals for teachers; professional schools, universities, and workshops specializing in pottery; local resources in Fiorida; and suppliers. (JH)

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AUTHORIZED COURSE OF INSTRUCTION FOR THE



ART EDUCATION CERAMIC TECHNOLOGY

6688.02

DIVISION OF INSTRUCTION-1971



CERAMIC TECHNOLOGY 6688.02

ART EDUCATION

Written by: Louis M. Marinaccio

for the

DIVISION OF INSTRUCTION Dade County Public Schools Miami, Florida 1972



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PREFACE

Learning has been referred to by many contemporary educators as a noun; but it is a verb experience-full of action and involvement, doing and being. At least it should be--and that kind of involved learning is what this course of study is all about.

The Quinmester Visual Arts Education Curriculum construct is a long range developmental effort directed towards providing a general education for learners in the aesthetically related art education field. accomplish this goal, instructional courses of study have been developed basically for teachers by teachers. Many Dade art specialists in various arts media have been recruited by the Art Office to write over 75 new and innovative courses of study in the area of art education. Educational specialists from the four corners of this land, along with aestheticians, social critics, and behavioral scientists have hailed the philosophy of the overall art curriculum construct undertaken by the Division of Instruction to be consistent with the latest national trends in art education, and to be an exemplary example of "success" oriented curricula designed to provide intense involvement in aesthetics and creative arts through group and individualized participation on the part of the learner.



All courses of study produced have been constructed with one major goal in mind: to provide a broad framework of goals and objectives; content; instructional procedures and strategies; and suggested learning activities. Many of the technically oriented courses of study list a variety of "work Sheets" designed to assist the learner with specific and highly technical studio procedures delineated in a manner so that art specialists (teachers) can use them "as is," or utilize the source information as a basis for producing "Learning Activities Packages." The appendix may include other pertinent material needed for today's contemporary art curriculum, e.g., vocabulary, resources for both learner and teacher, etc.

Constructive criticisms or recommendations relating to this publication are invited; please send to: Art Education Office, Room 300, Lindsey Hopkins, A-1.

Charles M. King, Consultant Art Education



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- I. COURSE TITLE
 CERAMIC TECHNOLOGY
- II. COURSE NUMBER
 6688.02

III. COURSE DESCRIPTION

A study of the preparation and development of clay bodies and ceramic glazes. Developing knowledge of properties of various ceramic materials; earth materials and metallic oxides - calculating batches and glaze formulas.

IV. RATIONALE

The art educator should foster the growth of what is unique in the individual and help him to develop a trong sense of empathy toward his visual environment; empathy toward the sensuous quality of the material; and empathy toward the content. There should be a kinesthetic interaction between the student and the media.

Emphasis will be placed upon preparing and developing aesthetic, functional, and suitable clay bodies and ceramic glazes.



V. COURSE ENROLLMENT GUIDELINES

- A. This course may be coordinated with the chemistry and math departments
- B. Prerequisite: Creative Ceramic Arts

VI. COURSE OF STUDY OBJECTIVES

- A. Competencies: The student upon completion of this unit will be able to:
 - 1. Definition and background
 - a. Exhibit a working knowledge of material, form, decoration, and glaze treatment of clay as created by the Egyptians
 - b. Analyze glaze aesthetically and chemically as viewed in the ceramic wares of the Sung Dynasty, the Ming Dynasty and the Ch' ing Dynasty
 - 2. Preparing and developing clays and clay bodies
 - a. Demonstrate several methods of adjusting clay bodies in order to improve their usefulness
 - b. List the basic characteristics of stoneware and porcelain clay bodies



- c. Identify several high-fired clay bodies
 - d. List the basic differences between earthenware clay and stoneware clay
- 3. Preparing and developing various ceramic glazes
 - a. Identify several types of early ceramic glazes
 - b. Demonstrate the ability to calculate from batch to formula
 - c. Create several series of reduction fired test tiles
 - d. List ten oxides frequently used in reduction firing

VII. COURSE CONTENT

- A. Definition and background
 - 1. Historical
 - a. Egyptian
 - b. Ancient Near East
 - c. Asia: (India, China, Korea, and Japan)

- d. Classical: (Crete, Greece, Etrusca, and Rome)
- e. Byzantium (Rome)
- f. Islam
- g. Pre-Columbian
- h. Renaissance in Europe
- 2. Contemporary
 - a. Aesthetic reaction
 - b. Craft Movement
 - c. American Indian
 - d. Ceramic Technologist
- B. Preparing and developing clays and clay bodies
 - 1. Methods of preparing and developing clay
 - a. Mixing clay
 - b. Wedging clay
 - c. Kneading clay
 - d. Conditioning clay
 - e. Aging clay
 - f. Storing clay
 - 2. Classification of clays
 - a. Residual or primary
 - b. Sedimentary or secondary

- 3. Kinds of clay
 - a. Kaolin
 - b. Ball clays
 - c. Fire clays
 - d. Sagger clay
 - e. Stoneware clay
 - f. Porcelain clay
 - g. Earthenware clays
 - h. Other kinds of clay
 - (1) Flint
 - (2) Shale
 - (3) Bentonite
 - (4) Terra Cotta
- 4. Characteristics and properties of clay and clay bodies
 - a. Plasticity
 - b. Water of plasticity
 - c. Firing range
 - d. Thermal reaction
 - e. Color
 - f. Deformation
 - g. Density
 - h. Porosity
 - i. Shrinkage

5. Clay Bodies

- a. Methods of altering and adjusting clay bodies to improve their usefulness
- Stoneware as an example of a high-fired clay body
- c. Porcelain as an example of a clay body
- d. Method of blending materials to formulate a suitable clay body
- e. Formulating clay bodies designed for specific methods of construction
- f. Designing clay bodies for throwing
- g. Designing clay bodies for modeling
- h. Designing clay bodies for casting
- i. Designing clay bodies for pressing and jiggering
- j. Color and texture in clay bodies
- k. Earthenware as a clay body
- 1. Stoneware as a clay body
- m. Porcelain as a clay body .
- C. Preparing and developing various ceramic glazes
 - 1. Characteristics and properties of glass and glazes
 - a. Silica as the basis of glass

- b. Creating glass
- c. The difference between glass and glaze

2. Types of early glazes

- a. Egyptian glazes
- b. Lead glazes
- c. Ash glazes
- d. Salt glazes
 - e. Slip glazes
 - f. Feldspathic glazes

3. Basic oxides and their function in glaze forming

- a. Oxidation and the oxides
- b. Glazs oxides
- c. Function of oxides in glazes
- d. The process of glaze melting in kiln

L. Ceramic Oxides

- a. Aluminum Oxide Al₂0₃
- b. Antimony Oxide SB203
- c. Barium Oxide BaO
- d. Boric Oxide B203
- e. Calcium Oxide CaO
- f. Chromium Oxide Cr₂O₃



- g. Cobalt Oxide CO203
- h. Copper Oxide CuO
- i. Iron Oxide Fe₂0₃
- j. Lead Oxide Pb0
- k. Lithium Oxide Li₂0
- 1. Magnesium Oxide MgO
- m. Manganese Dioxide MnO2
- n. Nickel Oxide NiO
- o. Potassium Oxide K20
- p. Silicon Dioxide SiO2
- q. Sodium Oxide Na₂0
- r. Strontium Oxide SrO
- s. Tin Oxide SnO
- t. Titanium Ti
- u. Vanadium V
- v. Zinc Oxide ZnO
- w. Zirconium Oxide ZrO2
- 5. Preparation of raw materials for use in glazes
 - a. Flint SiO2
 - b. Clay Al. 2032Si022H20
 - c. Feldspar
 - d. Whiting CaCO3
 - e. Magnesium Carbonate MgCO3
 - f. Dolomite CaCO3MgCO3



- g. Barium Carbonato Baco3
- h. Talc 3Mg0.4Si02.H20
- i. Strotium Carbonate SrCO3
- j. Colemanite 2Ca0.3B203.5H20
- k. Litharge Pb0
- 1. White Lead 2PbCO3Pb(OH)2
- m. Red Lead Pb304
- n. Zinc Oxide ZmO
- o. Antimony Oxide Sb₂O₃
- p. Soda Ash Na₂CO₃
- q. Sodium Carbonate Na₂Co₃
- r. Borse Ash, Calcium Phosphate 4Ca3 (PO4) 2.CaCo3.
- s. Cryolite Na3AlF6
- t. Lepidolite (HO₂F)₂KLiAl₂Si₃O₁₀
- u. Lithium Carbonate Li₂CO₃
- v. Fluorspar CaF_2 Pearl Ash, Potassium Carbonate $K_2^{CO}_3$
- w. Niter, Potassium Nitrate KNO3
- x. Borax Na₂B.2B₂O₃.10H₂O,
- 6. Theory and objectives of glaze calculations
 - a. Oxides are grouped according to their function in glazes
 - b. Theory of Atomic Weight
 - c. The Unity Formula



- d. Calculating Recipes of Glazes from their Formula
- e. Calculating from the batch to the formula
- 7. Calculating glaze using materials containing more than one oxide
 - a. Table of raw materials
 - b. Equivalent weight
 - c. Feldspar formulas
 - d. Calculation of the batch from the formula of a simple lead glaze
 - e. Calculation of glaze containing both feldspar and clay
 - f. Selection of raw materials for glaze
 - g. Calculation of several more complicated glazes
 - h. Calculation of glaze involving a complex feldspar
- 8. Calculating glaze formulas from recipes or batches
 - a. Determining the formula of a glaze from recipe
 - b. The process of calculating from the batch to the formula



- 9. Problems encountered in practical glaze calculation
 - a. Substitution of materials
 - b. Comparison If glazes
 - c. Alteration of glaze formulas
 - d. Experimentation of new makerials
- 10. Composition of glazes
 - a. Fusion points of glazes
 - b. Fluxing action of various oxides
 - c. The amounts of silica and alumina in glazes
 - d. Limit formulas
- ll. Types of glazes
 - a. Low-temperature alkaline glazes
 - b. Lead glazes
 - c. Glazes containing boron
 - d. Bristol glazes
 - e. Stoneware and porcelain glazes
- 12. Fritted glazes
 - a. Methods of preparing frits
 - b. Calculating a frit batch
 - c. The use of a frit as glaze



13. Glaze Textures

- a. Transparency
- }b. Opacity
 - c. Bright glazes
 - d. Mat glazes

14. Color in glazes

- a. Theory of color
- b. Oxides
 - (1) Iron oxide
 - (2) Copper oxide
 - (3) Cobalt oxide
 - (4) Manganese oxide
 - (5) Chromium oxide
 - (6) Nickel oxide
 - (7) Vanadium oxide
 - (8) Rutile
 - (9) Ilmenite
 - (10) Iron chromate
 - (11) Uranium oxide
 - (12) Cadmium
 - (13) Selanium
- c. Methods of blending, mixing, and compounding colored glazes
 - (1) Addition of single oxides to glazes

- (2) Line blends
- (3) Tri-axial blending
- (4) Use of glaze stains
- (5) Testing and analysis of glazes
- (6) Recording results in test glazes

15. Glaze mixing and application

- a. Ball milling
- b. Mixing and screening glazes
- o. Proportion of water in glazes
- d. Gums, floatatives, and flocculents for glazes
- e. Glazing by dipping, pouring, and painting
- f. Spraying glazes
- g. Thickness of glaze

16. Engobes

- a. Engobe composition
- b. Coloring oxides in engobes
- c. Vitreous engobes
- d. Methods of engobe application

17. Underglaze colors and decoration

- a. Pigments utilized in underglaze painting
- b. Hethods for applying underglaze
- c. Underglaze prints



- 18. Overglaze decoration
 - a. Majolica
 - b. Overglaze prints
 - c. Overglaze enamels
- 19. Special glaze effects and glazes
 - a. Crackle and pooled glazes
 - b. Egyptian raste
 - c. Red glazes
 - d. Terra sigillata
 - e. Raku ware
 - f. Luster
 - g. Salt glazing
 - h. Slip glazes
 - i. Ash glazes
 - j. Crystalline glazes
- 20. Reduction firing and reduction glazes
 - a. Theory of reduction firing
 - Results and effects of reduction on clay bodies
 - c. Reaction of base glazes to reduction
 - d. The color range of reduction glazes
 - e. Colors from copper oxide in reduced glazes
 - f. Colors obtained in reduction firing
 - g. Kiln instruction in reduction firing



VIII. COURSE PROCEDURES AND STRATEGIES, AND SUGGESTED LEARNING ACTIVITIES

A. Evaluation

It is essential to establish a criteria for evaluating the progress of the student in an art experience. Evaluation in ceramic art cannot be rigid to the extent that it will inhibit creative expression. Creativity is unique and personal.

The product itself cannot be evaluated without taking into consideration the process the student experienced from inception to completion. In addition, evaluation must include evidence of the growth of the individual in relation to his attitude, interest, ability to complete a project, how well he can use his past experience toward problem solving, respect for his own ability and the rights of others.

Evaluation is of vital importance to the student's development. It helps to determine the growth of the student so that the teacher can further motivate and guide the student toward his fullest self-development, creativity, and aesthetic growth.

- B. The criteria established for evaluation will vary due to individual differences among students and teachers. Each teacher must determine his own goals and formulate standards for evaluation always keeping in mind that evaluation must be positive as well as constructive. The following are some suggestions in setting up a criteria for evaluation:
 - l. Has the student learned to evaluate his own pottery as well as that of others with consideration to the sensuous quality of the clay, form, and content?
 - 2. Has the student designed the entire object with an awareness of space, form, movement, order, relationship of parts to the whole, and good color organization?



- 3. Has the student expressed his ideas creatively in the media in an original and meaningful way?
- 4. Has the student developed a sensitivity to the material?
- 5. Does the student express his ideas and individuality in clay?
- 6. Has the student become aware that texture results from an interaction of the clay and the tools?
- 7. Is the student aware of the difference between tactile and visual textures?
- 8. Has the student become sensitive to the expressive qualities of the different clay materials and tools?
- 9. Is the student aware that improper use of material and tools results in poorly constructed forms?
- 10. Is the student aware that variety can add interest to forms but too much can destroy it?
- 11. Does the student react empathically to clay in terms of three-dimensional forms?



- 12. Is the student familiar with good ceramic art of the past and present?
- 13. Is the student able to identify from contemporary ceramists the ways in which the craftsmen manipulate their tools and materials?
- 山. Has the student developed good work habits?
- 15. Has the student's behavior outside the art class improved as a result of his art experience?
- 16. Has the student developed a respect for his personal ability?
- 17. Has the student developed a respect for the rights of others?
- 18. Has the student acquired increased efficiency in handling materials and tools?
- 19. Has the student developed the ability to carry the project through to completion?
- 20. Has the student learned the firing process and how to use it to its fullest advantage?



- 21. Has the student developed good craftsmanship and yet retained the natural qualities of the clay?
- 22. Has the student learned to form pottery correctly so it does not warp or crack?
- 23. Is the product suited for the purpose for which it was made?
- 24. Does it incorporate the principles of good ceramic design?
- 25. Is the product the one best suited for work in clay?
- 26. Is the product well constructed?
 - 27. Does the product indicate individuality and expressive quality?
 - 28. Does the glaze fit the form?
 - 29. Has the student improved in attitude, interests, and development of technical skills?



IX. RESOURCES

- A. Suggested texts for pupils
 - Nelson, Glenn C., Ceramics, A Potter's Handbook. New York: Holt, Rinehart, Winston, 1971
 - Rhodes, Daniel, Clay and Glazes for the Artist Potter. Philadelphia: Chilton, 1957.

Suggested References

B. General

- Ball, F. Carlton and Lovoos, Janice, Making Pottery Without a Wheel. New York: Rheinhold, 1965.
- Hofsted, Jolyon, Step-by-Step Ceramics. New York: Golden Press, 1967.
- Kenny, John B., <u>Ceramic Sculpture</u>. New York: Greenberg, 1953
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C. Egyptian

Raphael, Max, Prehistoric Pottery and Civilizations in Egypt. New York: Phantom, Inc., 1947

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- D. Ancient Near East
 - Childs, V. Gordon, New Light on the Most Ancient
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 - Marinatos, S., Crete and Mycenae. New York: Harry N. Abrams, Inc., 1960
 - Richter, G. M. A., Attic Red-Figured Vases. New Haven: Yale Press, 1958.
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- G. Islam
 - Lane, Arthur, Early Islamic Pottery. London: Faber, 1939.
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- H. Pre-Columbian
 - Anton, Ferdinand and Dockstader, Frederick J.,

 Pre-Columbian Art and Later Indian Tribal

 Arts. New York: Harry N. Abrams, Inc., 1968
 - Bernal, Ignacio, 3000 Years of Art and Life in Mexico. New York: Harry N. Abrams, Inc., 1968.

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Westheim, Paul, The Sculpture of Ancient Mexico.

New York: Doubleday and Company, Inc.,
1963.

I. Renaissance in Europe

Goldscheider, Ludwig, Michelangelo's Models in Wax and Clay. London: Phaidon, 1962.

Honey, William B. Wedgewood Ware. London: Faber, 1949.

Liverani, Giusoppe, Five Centuries of Italian Majolica. New York: McGraw Hill, 1960.

Charleston, Robert J., World Ceramics. New York: McGraw Hill, 1968.

J. Suggested periodicals for pupils

Ceramics Monthly
Box 4548
Columbus, Ohio 43212

Craft Horizons 29 West 53rd Street New York, N.Y. 10019

School Arts 50 Portland Street Worcester, Mass. 01608

Design Quarterly 1710 Lyndale Avenue Minneapolis 3, Minnesota

K. Suggested places to visit

Grove House School of Art 3496 Main Highway Coconut Grove. Florida 33133

Sandpiper 2924 Florida Avenue Coconut Grove, Florida 33133

Village Corner Gallery 1136 South Dixie Highway Coral Gables, Florida

Lowe Art Museum 1301 Miller Drive Coral Gables, Florida

Miami Art Center 7867 North Kendall Drive Kendall, Florida

Ceramic League of Miami 7867 North Kendall Drive Kendall, Florida

Miami Museum of Modern Art 381 N. E. 20th Street Miami, Florida

Grove House Gallery 34,96 Main Highway Coconut Grove, Florida

Museum of Science-Planetarium 3280 South Miami Avenue Miami, Florida

Fairchild Tropical Garden 10901 Old Cutler Road Coral Gables, Florida

Fantastic Gardens 9550 S. W. 67th Avenue Miami, Florida

Miami Seaquarium Rickenbacker Causeway Virginia Key, Florîda Crandon Park Zoo Key Biscayne, Florida

Scholastic Art Awards Exhibition Burdine's Department Store 22 E. Flagler Street Miami, Florida

Miami Studio Shop 2363 W. Flagler Street Miami, Florida

Bass Museum of Art 2100 Collins Avenue Miami Beach, Florida

Japanese Gardens MacArthur Causeway Hiami, Florida

Vizcaya Art Museum 3251 South Miami Avenue Hiami, Florida

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XI. APPENDIX

SUGGESTED MEDIA RESOURCES FOR PUPILS AND TEACHERS

American Crafts Council Research and Education Dept. 29 West 53rd Street New York, N.Y. 10019

Scholastic Film Strips 906 Sylvan Avenue Englewood Cliffs, N.J. 07632 Contemporary Ceramics

A. Film strips

U.S.A., 1966

Forms from the Earth: 1000 Years of Pottery in America 1962

Art by Talented Toen-Agers - 1970: Sculpture and other Three-Dimensional Art

Art by Talented Teen-Agers - 1969: Sculpture and other Three-Dimensional Art

Art by Talented Teen-Agers-1968: Sculpture and other Three-Dimensional Art

Art by Talented Teen-Agers - 1967: Sculpture and other Three-Dimensional Art

Art by Talented Teen-Agers - 1961-1966: Sculpture and other Three-Dimensional Art

Collected Works of Teen-Age Art - Sculpture and other Three-Dimensional Art American Crafts Council Research and Education Dept. 29 West 53rd Street New York, N.Y. 10019

- B. Slides
- F6 Three Ceramists:
 Gronborg/Leady/
 Williams, 1969 Ceramic sculpture
 from one-man
 exhibitions at M.C.C.
- C14 Young Americans, 1969 Winning entries in all craft media from the national competition of this title sponsored by the American Crafts Council.
 - C9 Objects Are ...?, 1968
 An exhibition which sought to question the nature of objects.
 Covers work by 42 craftsmen in various media.
- C7 Ceramics of Maija Grotell, 1968 Pottery by the head of the Department of Ceramics at Cranbrook Academy of Art from 1930-1966.
- Bly Craftsmen U.S.A. '66 National Merit Awards in a competition sponsored by the American Crafts Council; covers all craft media.
- Bl The American Craftsmen, 1964 - 30 craftsmen and their work in clay, metal, wood, and textiles.
- All4 Forms from the Earth,
 1000 Years of Pottery
 in America, 1962 A
 summary view of pottery
 in America up to 1961.

Slides

- Al3 Young Americans, 1962 Winning entries in all craft media from the national competition of this title sponsored by the American Crafts Council.
- A6a <u>Katherine Choy, 1961-</u>
 Pottery from a retrospective exhibition
 at M.C.C.
- Fil General Ceramics, 1960 Functional and sculp-tured ceramics by contemporary American craftsmen.
- D5 Contemporary Ceranics U.S.A., 1966 - Works by 77 American Craftsmen.
- D6 General Ceramics, 1963 Work of the early 1960's by American craftsmen.
- Fly Objects U.S.A., 1969 A selection of work in all media from the Johnson Collection of Contemporary Grafts.
- Bl2 Folk Art of North India,

 1966 Ceramics, embroidery,
 applique, and jewelry from
 North India.
- Clo Salt Glaze Hiln: building/
 firing, 1969 Construction
 and firing at the Byron
 Temple Studio; Script gives
 additional information on
 materials and techniques.
- Paul Soldner photographed as he fires pottery by the raku process

Prothmann Associates, Inc. 2795 Milburn Avenue Baldwin, New York 11510

Contemporary Ceramics - a survey 1946-1964 from prize-winning entries in National and International Exhibits at the Everson Museum of Art. Syracuse, New York.

International Ceramic Exhibitions, 1958.

C. Portfolios

Throwing a Cylinder

Professional Publications, Inc. Box 4548 Columbus, Ohio 43212

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Indianapolis, Indiana 46227

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The Nature of Clay

Earth, Fingers, and Fire

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Audio-Visual Center Indiana University Bloomington, Indiana Craftsmanship in Clay:

Simple Slab Methods

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Terry Weinberger

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Ceramic Age 9 Chester Building Cleveland 14, Ohio



Ceramic Industry 5 S. Wabash Avenue, Chicago 3, Illinois

Ceramic Data Book Industrial Publications 5 S. Wabash Avenue Chicago 3. Illinois

Ceramics Monthly Box 4548 Columbus, Ohio

Journal of the American Ceramic Society Columbus, Ohio

Journal of the National Council on Education for the Ceramic Arts Room 132 Fine Arts Building University of Illinois Champaign, Illinois

Journal of the National Art Education Association 1201 Sixteenth Street N.W., Washington, D. C.

Craft Horizons
29 West 53rd Street
New York, N.Y.

School Arts 50 Portland Street Worcester, Massachusetts

Design Quarterly 1710 Lyndale Avenue Minneapolis 3, Minnesota

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MAT CONE 017 White lead 27.5 3.9 Dolomite 12.6 Barium carbonate Oxford spar 7.4 11.5 China clay 36.9 Flint MEDIUM FIRE-MAT CONE 04 Feldspar 40 6 Whiting Barium carbonate Zinc oxide Talc Kaolin Ball clay Flint CONE GLAZE 15.3 Whiting 23 White lead 30 Feldspar Kaolin Flint ZINC CRISTAL CONE 04 Soda ash 13.1 Boric acid 15.2 12.7 Zinc oxide Flint 41.8 Rutile 6.6 6 Ball clay CONE OF AVENTURINE 99 Ferro frit 3304 Red iron oxide 16 5 Florida kaolin LEAD COLEMANITE CONE OF 7.2 Flint 52.9 19.4 Oxford spar White lead 5.5 Dolomite 12.4 Colemanite



Zinc oxide

2.4

COLEMANITE CONE OF 46.8 Oxford spar Flint 20 2.5 China clay Whiting 13.7 Colemanite 4 Zinc oxide Dolomite MILKY COLEMANITE COME 43.9 Oxford spar 24.7 Flint Clay 1 20.6 Colemanite 1.5 Whiting 3.2 Zinc Oxide Barium carbonate .. 6.4 TRANSPARENT COLEMANITE CONE OL 47 Oxford spar Colemanite 17.4 Barium carbonate .. 7.2 TRANSPARENT GREEN CONE OL White lead 30 Whiting 111.2 Feldspar 12.9 Kaolin 48 Flint 9.48 Copper Oxide CONE OF TRANSPARENT BLUE 116.1 White lead 30 Whiting 111.2 Feldspar Kaolin 12.9 48 Flint 4 Cobalt oxide CONE O4 CLEAR LEAD White lead Oxford spar 31.3 3 China clay 1.8 Talc 11.8 Flint 6 Whiting



BARIUM MAT CONE	Ol ₁
White lead	43.2 11 39.2 2.1
GREEN ANTIQUE CONE	03-04
White lead	41.3 48.24
BLUE ANTIQUE CONE	03-04
White lead	237.37 3.5 4.3 1.9
CREAM ANTIQUE CONE	03-04
White lead	232.2 49.8 49.5 7 16
HIGH LIME, SEHI OPAQUE	CONE 04
White lead Oxford spar Flint China clay Whiting Magnesium carbonate	5.4
TRANSPARENT LEAD CON	E 04
White lead Oxford spar Flint China clay Whiting	52.1 20.3 18.7 3.0 5.8



WHITE OPAQUE CONE 05-04 White lead 129 Whiting 30 Canadian feldspar ... 111.4 12.9 Kaolin 48 Flint 45 Tin oxide ORANGE OPAQUE CONE 05-04 129 White lead Whiting 30 111.4 Canadian feldspar . 12.9 Kaolin 48 Flint Tin oxide Iron oxide 19.2 CONE 05-04 BROWN OPAQUE White lead 141.9 30 Whiting Eureka spar 112.8 Kaolin 10.3 Flint 43.4 30 Tin oxide Chestnut brown base 30 GREEN RUTILE MAT CONE 04 227.8 White lead 41.3 Feldspar 6.45 Kaolin 48.24 Flint 22.8 Rutile Copper oxide YELLOW TRANSPARENT CONE 04-05 129 White lead 30 Whiting 111.2 Canadian feldspar . 12.9 Kaolin 48 Flint 33 Yellow base



BLUE TRANSPARENT CONE 04-05
White lead
TRANSPARENT PURPLE CONE 04
White lead
TRANSPARENT OCRE CONE OL
White lead
TRANSPARENT WHITE CONE O4
White lead
Ferro-enamel frit 3304 80 Ball clay 5 Red iron oxide 15
IRON RED CONE 04
Ferro-enamel frit 3304 87 Ball clay 5



BROWN LUSTRE CONE 04	
White lead	58 13 4 16 3
COLEMANITE BARIUM MAT CON	E 04
Oxford spar Flint Colemanite Whiting Barium carbonate Zinc oxide RETICULATION CONE 04	46.8 10 11 13 13
Ferro-enamel frit 3304 Colemanite Ball clay Zircopax	68 12 8 12
LUSTRE COME OL	
Ferro-enamel frit 3195 Ball clay Silver nitrate Bismuth Subnitrate	87 10 1 2



BASIC GLAZE CONE O	<u>5</u>
White lead Whiting Feldspar Kaolin Flint	100 15 35 8 25
RAW ALKALINE CONE O	5
Borax Soda Ash Whiting Feldspar China clay Flint	32 16 7 20 5 20
CLEAR GLAZE CONE O	5
Frit #k-3	20 4 2•5 2
ALKALINE CONE 05	
Soda ash	64 30 56 42
VOLCANIC ASH CONE O	5
Volcanic ash Colemanite Bentonite	67 28 5
STONEWARE CONE 05	
Feldspar Dolomite Whiting China clay Flint	35 16 6 9 34



LOW FIRE TRANSPARENT COME 05
White lead
JEWEL GLAZE CONE 05
Pemeo frit P-64
TRANSPAREIT BLUE GLAZE COHE 05
White lead 116.1 Whiting 30 Canadian feldspar 111.2 Kaolin 12.9 Flint 48
GUN METAL BLACK CONE 05
Soda ash
AVENTURINE CONE 05
Borax
BOROSILICATE GLAZE CONE 05
Colemanite 30 Feldspar 45 Zinc oxide 5 Barium carbonate 6 China clay 4 Flint 10



LITHIUM BLUE COME 05	
Lithium carbonate	13.6 53.9 2.8
OPAQUE WHITE GLAZE CONE White lead Whiting Canadian feldspar Kaolin Flint Tin oxide	129 30 111.4 12.9 48 45
MAT OPAQUE CONE 05 Lead carbonato Feldspar EPK China clay Silica Rutile BARIUM MAT CONE 05	610 110 30 190 60
White lead Whiting Feldspar Barium carbonate Calcined Zinc oxide Flint China clay PURPLE GLAZE CONE 05	38 19 13 15 7
Soda Ash Magnesium carbonate Boric acid Flint Cobalt oxide	50 25 200 50 5
YELLOW MAT COME 05 White lead Whiting Eureka spar Kaolin Georgia clay Yellow base	155 30 83 33 13 28



CONE 05 GREEN MAT White lead 124 Whiting 30 Eureka spar 33 Kaolin Georgia clay Copper exide BLUE HAT COME 05 121.6 White lead 20 Whiting Calcined kaolin 13.9 Barium carbonate 39.4 55.7 China clay 31.02 Flint COME 05 WHITE MAJOLICA MAT Frit #3304 70 Ball clay Tin oxide 23 WHITE MAT CONE 05 155 30 White lead Whiting Eureka spar Calcined kaolin Georgia clay BROWN MAT COME 05 White lead 129 30 83 Whiting Eureka spar 33 13 Calcined kaolin Georgia clay Manganese carbonate ... PINK MAT CONE 05 155 30 White Whiting Eureka spar Calcined kaolin Georgia clay Pink base



BUTTERSCOTCH OPAQUE CO	NE 06
Barnard slip Wood ash Lithium carbonate Bentonite	45 45 10 2
MATT PATELINED GLAZE CO	ME 06
Lepidolite	50.3 19.9 29.8
FOAM OR LAVA GLAZE CO	-
O. Hommel Frit Tin oxide Bentonite Barium sulfate JEWEL GLAZE CONE 06	14 1.5 1 30
Pemeo frit	<u>2</u> 8
BASIC CLEAR TRANSPARENT C	ONE OC
White lead Whiting Feldspar Kaolin Flint	15
CELADONE REDUCTION CO	NE 06
Potash feldspar Whiting Kaolin Flint Red iron oxide	61.3 7.3 4.9 24.8
CHOCOLATE CONE 06	
Albany slip	85 11 4 2



<u>co</u>	PPER	HED F	REDUC	TTON	C	ONE	06	
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MATT GLAZE CONE 06 18.8 Zinc oxide 3.9 8.6 Whiting Lithium carbonate ... China clay 21.1 41.5 Flint Titanium oxide 6.1 **COME 06** ALUMINA MATT White lead 40 11 Whiting Feldspar 20 China clay 16 Flint 7 PORCELAIN GLAZE GONE 06 27 Feldspar 20 Whiting China clay 20 33 Flint ALUMINA COLEMANITE CONE 06 White lead 34 15 Whiting Colemanite 10 9 Feldspar China clay 14. 17 Flint 11 Zereopax SEMINAT GLAZE CONE 06 9.8 Whiting 2.1 Talc Barium carbonate 11 4.3 Calcined zinc oxide . Soda feldspar 8.3 Kaolin Ball clay 8 Flint 14



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	Feldsp Colomi Whitin China Flint	te . g .a clay	• • • • •		• • • • •	35 16 6 39 34



CONE 05 COLEMANITE GLAZE Colemanite Feldspar Zinc oxide Barium carbonate China clay 10 Flint ALKALITIE GLOSS GLAZE CONE 05 28 Soda ash 10 Whiting 50 Feldspar 12 Flint BRISTOL GLAZE USING DOLOMITE CONE 05 6 Zinc oxide Dolomite 10 67 Feldspar 8 China clay 9 Flint CONE 05 MAT (RAW OR DISQUE WARE) 52...52 Foldspar 6...8 Silica 20...20 Whiting 9... 9 Kaolin 10...16 Bentonite Zircopax

Oxidation. Reduction

